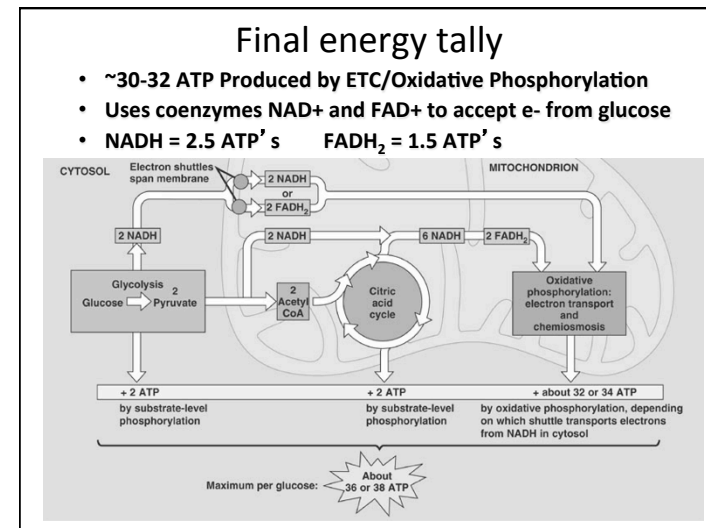
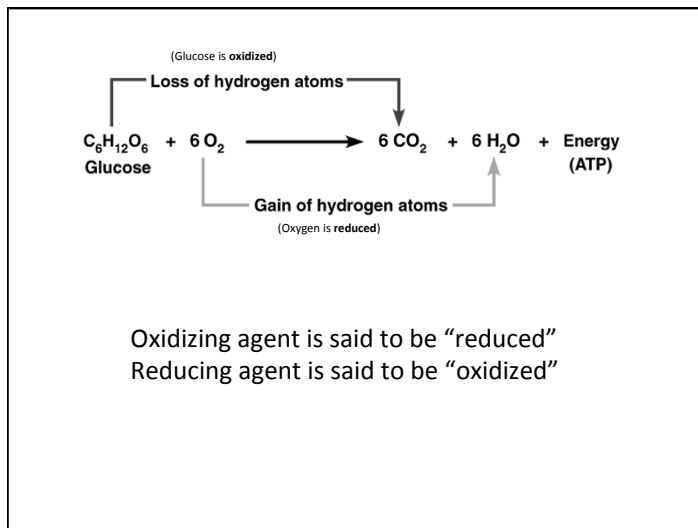
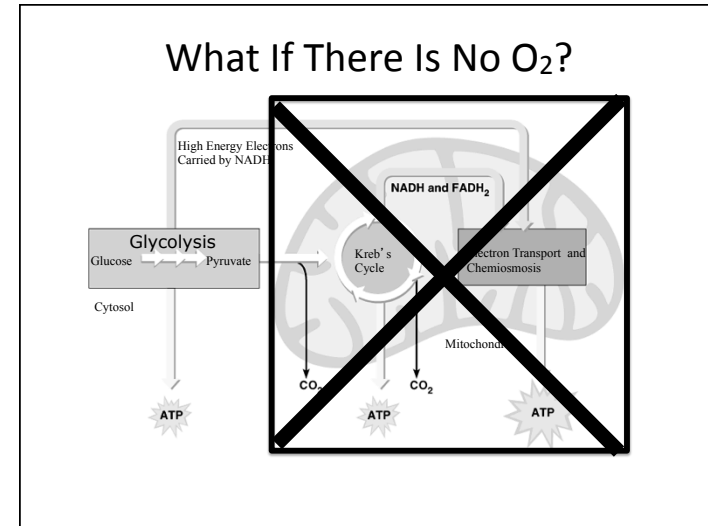
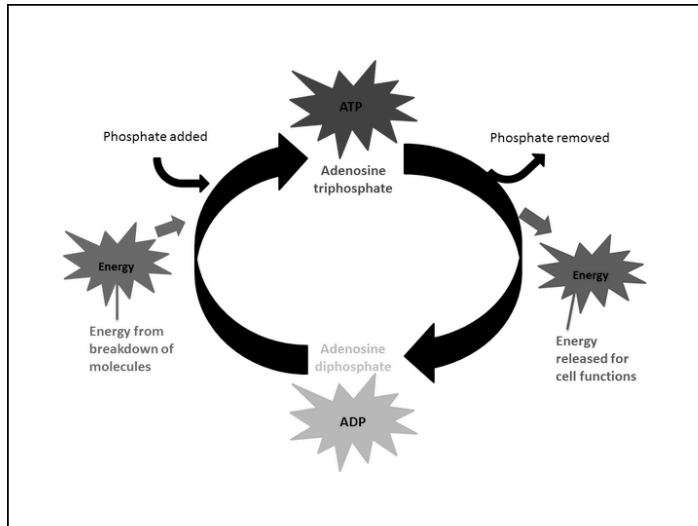


Topics

- Cellular Respiration
- Fermentation
- Energy sources
- Exam Review

Step	Input	Output	Important Features
Glycolysis	<ul style="list-style-type: none"> Glucose ADP, P NAD⁺ 	<ul style="list-style-type: none"> Pyruvic acid 2 ATP Electrons and H⁺ carried by NADH 	<ul style="list-style-type: none"> Occurs in cytoplasm All organisms use this pathway NO O₂ required ATP is produced by direct phosphate transfer
Citric Acid (Krebs) Cycle	<ul style="list-style-type: none"> Pyruvic acid ADP, P NAD⁺, FAD 	<ul style="list-style-type: none"> CO₂ 2 ATP Electrons and H⁺ carried by NADH and FADH₂ 	<ul style="list-style-type: none"> Occurs in mitochondria (central matrix) Pyruvic acid first converted to Acetyl CoA Other fuel molecules (fatty acids, amino acids) can be used for Citric Acid Cycle Carbon atoms of original glucose completely converted to CO₂ ATP is produced by direct phosphate transfer
Electron Transport Chain (E.T.C.)	<ul style="list-style-type: none"> O₂ Electrons and H⁺ carried by NADH and FADH₂ from glucose ADP, P 	<ul style="list-style-type: none"> H₂O 34 ATP NAD⁺ and FAD recycled 	<ul style="list-style-type: none"> Occurs in mitochondria (inner membrane) Oxygen is final electron acceptor and, with H⁺, forms H₂O Energy of electrons is used by proteins of E.T.C. to pump H⁺ across inner membrane of mitochondrion to generate H⁺ concentration gradient and potential energy ATP synthase uses H⁺ gradient to make ATP (like turbine) by chemi-osmosis



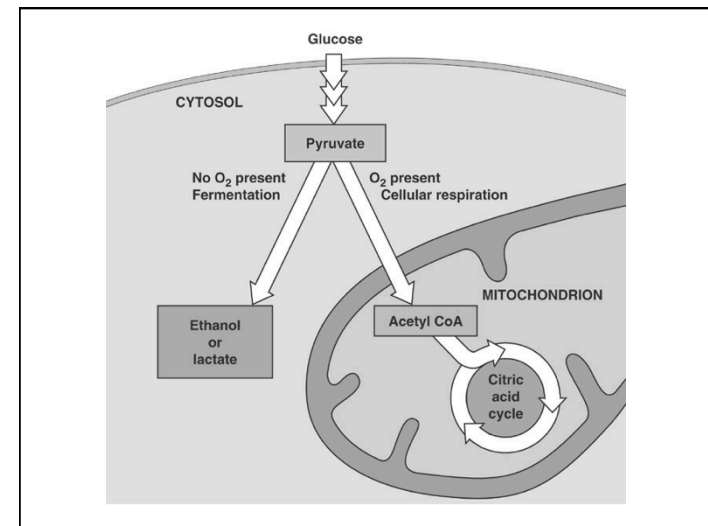


Anaerobic Cellular Respiration

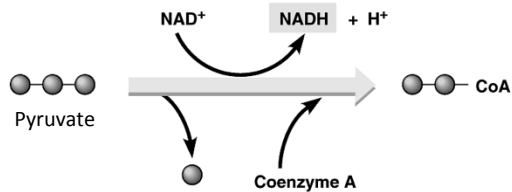
- Some organisms thrive in environments with little or no oxygen
 - Marshes, bogs, gut of animals
- No oxygen used = 'an' aerobic**
- Final electron acceptors differ, eg. sulfate and nitrate
- Less efficient at making ATP than aerobic respiration



VS. fermentation



In Aerobic Conditions



What if there is no O₂?...

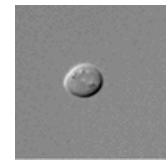
Without oxygen the cell cannot convert pyruvate to Acetyl-CoA because it cannot oxidize NADH back to NAD⁺

...pyruvate and NADH build up!!

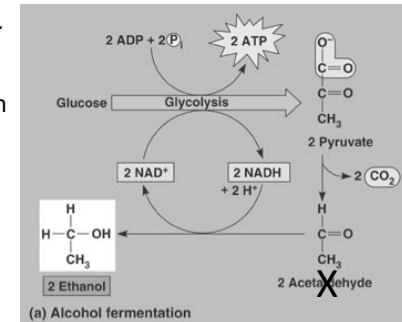
Fermentation

What if there is no O₂?...

1) Alcoholic fermentation Yeast Bacteria



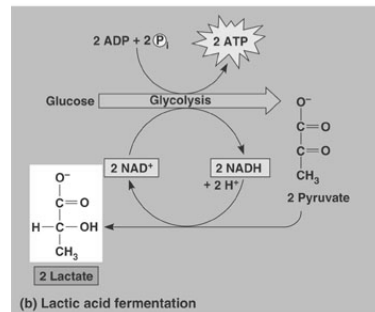
Saccharomyces cerevisiae



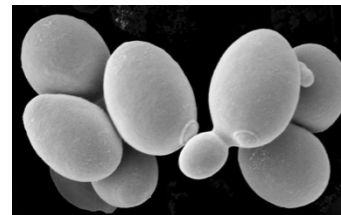
Fermentation

What if there is no O₂?...

2) Lactic acid fermentation Bacteria Animal cells (muscle)



Why use Fermentation?



Saccharomyces cerevisiae

Yeast are called *facultative anaerobic* organisms (they can do both **aerobic respiration** and **fermentation**)

Pros

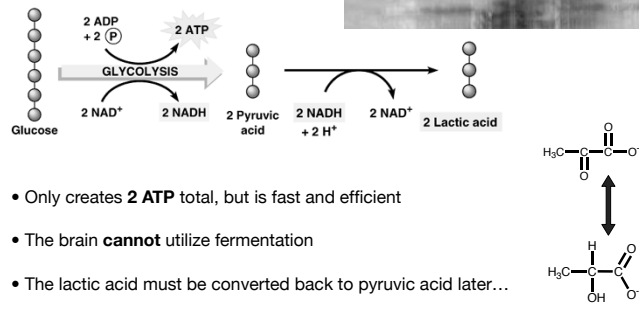
- Fermentation requires no oxygen
- Fermentation is MUCH faster than oxidative phosphorylation

Cons

- Only creates **2 ATP** total
- Creates waste that is more difficult to remove (and toxic) than just water and CO₂

Fermentation

What about in humans?

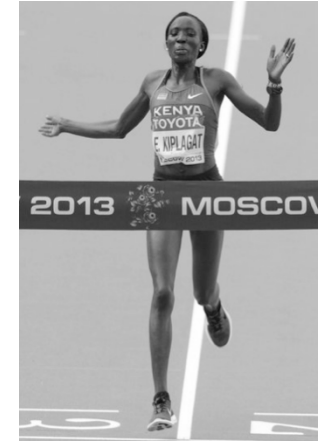


- Only creates **2 ATP** total, but is fast and efficient
- The brain **cannot** utilize fermentation
- The lactic acid must be converted back to pyruvic acid later...

Fast Twitch

VS.

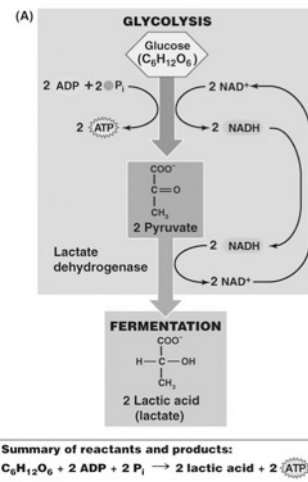
Slow Twitch



Fermentation

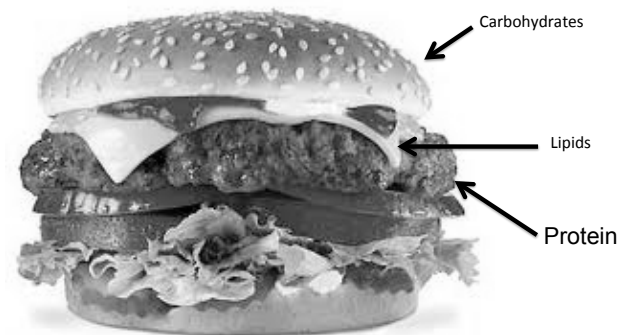
What if there is no O_2 ?

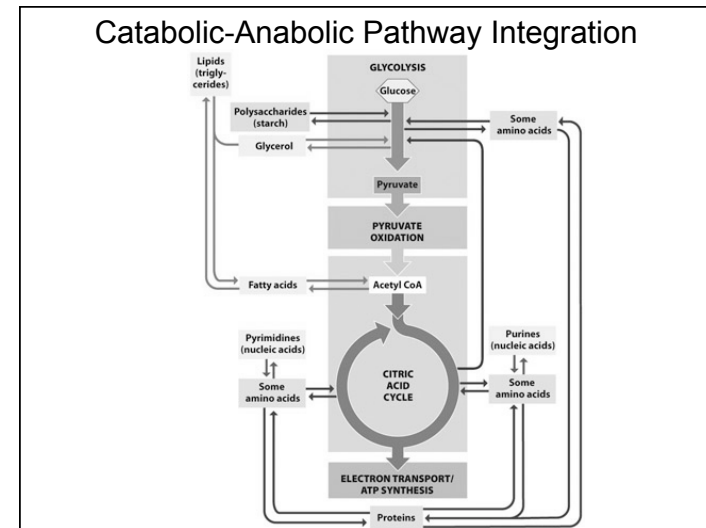
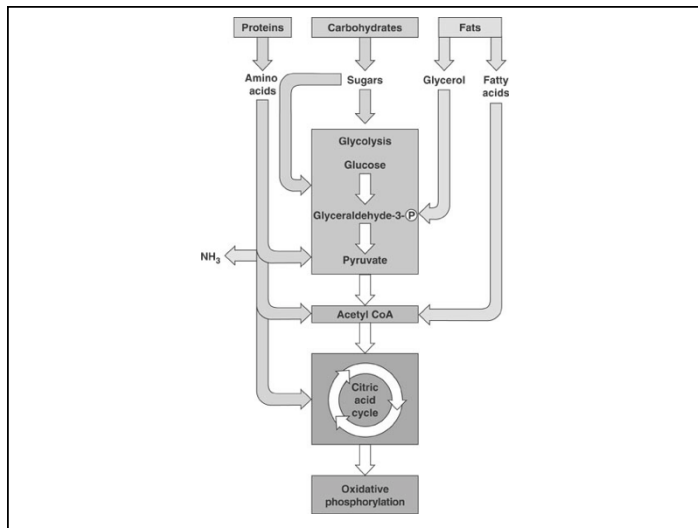
- 1) Lactic acid fermentation
Bacteria
Animal cells (muscle)



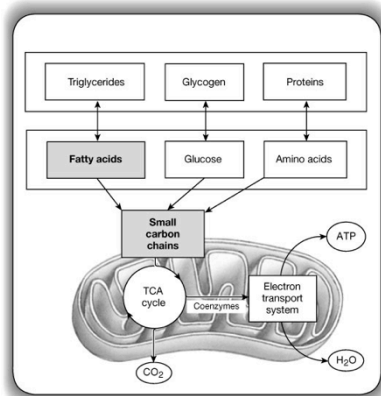
PRINCIPLES OF LIFE, Figure 6.13
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Catabolic interconversion



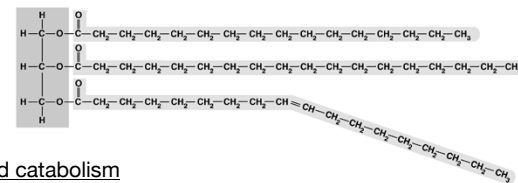


Lipid Catabolism



Lipid Catabolism

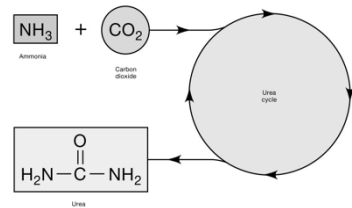
- **Lipolysis** breaks lipids down to components
- **-triglycerides** broken down to **fatty acids** and **glycerol**
glycerol is further broken down and used in glycolysis
Enters the citric acid cycle as **Acetyl-CoA**



Fatty acid catabolism

- Takes place in the mitochondria and/or peroxisomes
- Fatty acids are **highly reduced**. One average fatty acid can yield more ATP than 4.5 glucose molecules!

The Urea Cycle



- NH_4 (from amino group) is very toxic, urea is fairly harmless and very water soluble
- Takes place in the **Liver and Kidneys**
- Requires energy